

New Synthesis of Dialkyl Fluoroalkynylphosphonates

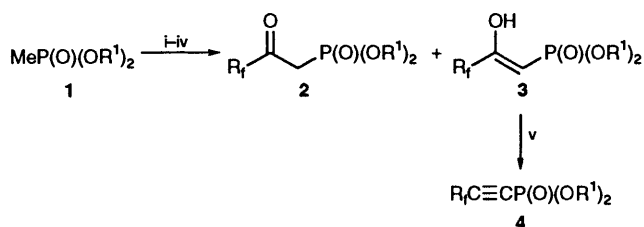
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A new synthesis of dialkyl fluoroalkynylphosphonates involving trifluoromethanesulfonation and elimination starting from commercially available substances is described.

Many phosphonic acids and their derivatives have been shown to exhibit biological activities.¹ Vinylphosphonates have attracted much interest from synthetic organic chemists.^{2,3} Considerable efforts have recently been made to develop methodologies for the synthesis of fluorinated organic compounds owing to their enhanced biological and physiological activities and fluorine-containing vinylphosphonates are regarded as useful intermediates in organic synthesis.² However, to the best of our knowledge, methods for the preparation of fluoroalkynylphosphonates are still limited. Diphenyl fluoroalkynylphosphonates have been prepared by intramolecular Wittig reaction,⁴ however, the attempted preparation of dialkyl fluoroalkynylphosphonates by this methodology failed as it was not possible to prepare the corresponding dialkoxyphosphorylmethylene(triphenyl)phosphorane.⁵ We now report a new synthetic method for the preparation of dialkyl fluoroalkynylphosphonates in 65–80% yields starting from commercially available substances *via* trifluoromethanesulfonation and elimination.

Lithium dialkyl methylphosphonate generated from **1** and butyllithium was treated with fluoroalkanoate **5**, to give, after distillation, a mixture of **2** and **3**,⁶ which can be converted into the fluoroalkynylphosphonate by treatment with trifluoromethanesulfonic anhydride in the presence of *N,N*-diisopropylethylamine as shown in Scheme 1. The results are summarized in Table 1.



Scheme 1 Reagents and conditions: i, BuLi–THF, -78°C ; ii, $\text{R}_f\text{CO}_2\text{R}^2$, $5, < -50^{\circ}\text{C}$; iii, $6 \text{ mol dm}^{-3} \text{ HCl}$, -20°C ; iv, distillation; v, $(\text{CF}_3\text{SO}_2)_2\text{O}$, Pr^1_2NEt , CH_2Cl_2 , 0°C

Experimental

The mixture of **2** and **3** was prepared according to the method reported by Nickson.⁶

General Procedure for the Preparation of Fluoroalkynylphosphonates 4.—Into a solution of a mixture of **2** and **3** (3 mmol in 3 cm^3 of dry CH_2Cl_2), diisopropylethylamine (10.8 mmol) was added at 0°C under nitrogen, followed by trifluoromethanesulfonic anhydride (3.6 mmol). After being stirred at 0°C for 4 h, the reaction mixture was diluted with diethyl ether (30 cm^3) and filtered, the residue was washed with diethyl ether ($2 \times 10 \text{ cm}^3$). The filtrate was collected, washed with water (10 cm^3), HCl (1 mol dm^{-3} ; 15 cm^3) and dried. Evaporation of the solvent gave a residue which was purified by flash column chromatography to give the product **4**.

Diethyl 3,3,3-trifluoroprop-1-ynylphosphonate 4a. B.p. $53^{\circ}\text{C}/10 \text{ mmHg}$ (Found: C, 36.35; H, 4.3. $\text{C}_7\text{H}_{10}\text{F}_3\text{O}_3\text{P}$ requires C, 36.55; H, 4.35%; $\nu(\text{film, neat})/\text{cm}^{-1}$ 2986s, 2220w, 1253s and 1023s; $\delta_{\text{H}}(\text{CDCl}_3/\text{internal SiMe}_4)\dagger$ 4.27 (m, 4 H) and 1.42 (t, J 7.0, 6 H); $\delta_{\text{F}}(\text{CDCl}_3/\text{external trifluoroacetic acid, positive for downfield shifts}$ 26.1 (s, 3 F); m/z 231 (M + 1), 185, 161 and 93.

Diisopropyl 3,3,3-trifluoroprop-1-ynylphosphonate 4b. B.p. $56^{\circ}\text{C}/6 \text{ mmHg}$ (Found: C, 41.7; H, 5.6. $\text{C}_9\text{H}_{14}\text{F}_3\text{O}_3\text{P}$ requires C, 41.9; H, 5.4%; ν/cm^{-1} 2982s, 2220w and 1254s; δ_{H} 1.40 (d, J 7.0, 12 H) and 4.60–5.10 (m, 2 H); δ_{F} 25.8 (s, 3 F); m/z 259 (M + 1), 201 and 189.

Diethyl 3-chloro-3,3-difluoroprop-1-ynylphosphonate 4c. B.p. $68^{\circ}\text{C}/7 \text{ mmHg}$ (Found: C, 33.8; H, 3.9. $\text{C}_7\text{H}_{10}\text{ClF}_2\text{O}_3\text{P}$ requires C, 34.1; H, 4.0%). ν/cm^{-1} 2985s, 2220w, 1279s and 1023s; δ_{H} 1.28 (t, J , 7.0, 6 H) and 3.92–4.40 (m, 4 H); δ_{F} 37.5 (s, 2 F); m/z 247 (M + 1), 211, 202 and 183.

Diisopropyl 3-chloro-3,3-difluoroprop-1-ynylphosphonate 4d. B.p. $75^{\circ}\text{C}/7 \text{ mmHg}$ (Found: C, 39.05; H, 5.15. $\text{C}_9\text{H}_{14}\text{ClF}_2\text{O}_3\text{P}$ requires C, 39.3; H, 5.1%; ν/cm^{-1} 2982s, 2220w, 1278s and 1001s; δ_{H} 1.40 (d, J 7.0, 12 H) and 4.55–5.10 (m, 2-H); δ_{F} 37.5 (s, 2 F); m/z 275 (M + 1), 256, 232 and 189.

Dimethyl 3-chloro-3,3-difluoroprop-1-ynylphosphonate 4e. B.p. $55^{\circ}\text{C}/10 \text{ mmHg}$ (Found: C, 27.1; H, 2.8. $\text{C}_5\text{H}_6\text{ClF}_2\text{O}_3\text{P}$ requires C, 27.4; H, 2.7%). ν/cm^{-1} 2982s, 2220w, 1278s and 1021s; δ_{H} 3.88 (d, J 12.5, 6 H); δ_{F} 36.7 (s, 2 F); m/z 219 (M + 1), 217, 188 and 133.

Diethyl 3,3,4,4,5,5,5-heptafluoropent-1-ynylphosphonate 4f. B.p. $70^{\circ}\text{C}/9 \text{ mmHg}$ (Found: C, 32.4; H, 3.0. $\text{C}_9\text{H}_{10}\text{F}_7\text{O}_3\text{P}$ requires C, 32.7; H, 3.0%). ν/cm^{-1} 2987s, 2220w, 1234s and

$\dagger J$ Values are given in Hz.

Table 1 Preparation of dialkyl fluoroalkynylphosphonates **4**

Compound	R _f	R ¹	R ²	Yields of 2 + 3 (%) ^a	2 : 3 ^b	Yields of 4 ^c (%) ^a
a	CF ₃	Et	Et	92	86:14	80
b	CF ₃	Pr ⁱ	Et	81	82:18	75
c	ClCF ₂	Et	Me	83	78:22	70
d	ClCF ₂	Pr ⁱ	Et	88	75:25	65
e	ClCF ₂	Me	Me	85	75:25	65
f	n-C ₃ F ₇	Et	Et	92	80:20	75

^a Isolated yields. ^b The mixture of **2** and **3** was characterized by ¹H NMR, ¹⁹F NMR and IR spectroscopy. The ratios of **2**:**3** were estimated on the basis of ¹⁹F NMR spectroscopy. ^c All compounds **4** are new and characterized on the basis of microanalysis, IR and NMR spectroscopy and mass spectrometry.

1024s; δ_{H} 1.40 (t, J 7.0, 6 H), 4.02–4.50 (m, 4 H); δ_{F} –2.2 (s, 3 F), –24.1 (s, 2 F), –49.1 (s, 2 F); m/z 331 ($M + 1$), 285 and 261.

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